A Tactically Novel Alternative to Acyclic Stereoselection Based on the Concept of a Replicating Chiron -1,3- and 1,4-C-Methyl Substitution[†]

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<u>Summary</u> - A chiral 4-hydroxymethyl butenolide is used as a template for the stereocontrolled conjugate addition of a C-methyl group. A sequential two-carbon extension and reformation of a lactone with the new side-chain leads to a "replicated" butenolide, which is subjected to a second conjugate addition. This process leads to a seven-carbon chain with a predictable 1,3-C-methyl substitution pattern.

In the preceeding paper¹, we demonstrated the utility of a chiral butyrolactone derivative prepared from (S)-glutamic acid as a template for the stereocontrolled introduction of C-methyl substituents having a 1,5-relative disposition on a seven-carbon framework. The operational basis for such a strategy was predicated upon the concept of a replicating chiron, generated as a result of the inherent symmetrical disposition of a pivotal functional group (hydroxyl) with regard to other flanking substituents.

Scheme 1



 $^{^{\}dagger}$ Dedicated to Prof. H.H. Wasserman on the occasion of his 65th birthday.

In this paper we show the flexibility of this approach in the construction of acyclic carbon chains bearing C-methyl substituents with 1,3- and 1,4- disposition, with and without an intervening hydroxyl group. Such patterns of substitution are present in the segments of a number of natural products of relevance as well as in the structures of an important class of enzyme inhibitors.² Several studies have addressed this problem and a number of innovative approaches have been reported.³

Based on previous experience⁴ and our own studies, it was reasoned that the general retrosynthetic analysis shown in Scheme 1 might be successfully realized, leading to several combinations of 1,3-situated C-methyl substituents on a seven carbon framework. Thus, stereocontrolled conjugate addition, chain-extension, and repetition of the cycle on a replicated butenolide-type chiron would be expected to provide a chiral butyrolactone with two C-methyl substituents flanking the pivotal ring oxygen. These operations were successfully accomplished with virtually complete stereoselection.

The butenolide 2, mp 83-84°, $[\alpha]_D$ -83.2°,⁵ obtained from the known lactone 1^6 or from D-ribonolactone⁷ (Scheme 2), was treated with lithium dimethyl cuprate to give the C-methyl derivative 3, $[\alpha]_D$ +30.5° in 87% yield. Transformation of 3 via acyclic intermediates led to the epoxide 5, mp 62-63°, $[\alpha]_D$ -9°, in excellent overall yield.



a. t-butyldiphenylsilyl chloride, CH_2Cl_2 pyr., DMAP, 18h, (85%); b. LiN(TMS)₂, THF, 20 min.; then PhSeCl; c. H_2O_2 , CH_2Cl_2 , 0°, 30 min., (50%, 2 steps); d. Me_2CuLi , Et_2O_2 , -20°, 30 min., (87%); e. BH₃.Me₂S, THF, 18h, (quant.); f. Ph₃CCl, pyr., DMAP, 18h, (quant.); f. Ph₃CCl, pyr., DMAP, 18h, (92%); g. (n-Bu)₄, N⁺F⁻, THF, 2h, (92%); h. 2,4,6-triisopropyl benzenesulphonyl chloride, pyr., 0°, 24h; i. NaOMe, MeOH, 0°, 30 min., (75%, 2 steps); j. PhSCH₂CO₂H, LiN(TMS)₂, THF, 0° to 25°; add epoxide, 18h; k. ethyl (dimethylaminoethyl) carbodiimide HCl, DMAP, Et_2O , (85%, 2 steps); l. mCPBA, CH_2Cl_2 , -23°, 30 min.; m. toluene, CaCO₃, reflux, 45 min., (81%, 2 steps); n. LAH, Et_2O , 1h, (94%); o. MsCl, pyr., CH_2Cl_2 , then repeat n, (73%, 2 steps). Chain-extension¹ with dilithic phenylthicacetic acid, ⁸ followed by lactonization, oxidation and elimination gave the replicated chiral butenolide 7, $[\alpha]_D + 54.5^\circ$. With the bulky side-chain effectively shielding the α -face of the α,β -unsaturated lactone 7. A second cuprate addition gave the crystalline chiral lactone 8, mp 122-123°; $[\alpha]_D - 34.3^\circ$ as the sole product in 93% yield. Now that the cyclic lactone templates had served their purpose, acyclic equivalent structures could be considered. Thus, reduction of 8, followed by selective substitution of the primary hydroxyl group led to 9, $[\alpha]_D - 4.4^\circ$. Deoxygenation was easily achieved by reductive desulfonylation of the corresponding mesylate to give the (3R,5S)-dimethyl-1,7-heptanediol derivative 10, $[\alpha]_D - 3.6^\circ$, in high yield.

Since the conceptual basis of our strategy is predicated upon the exploitation of the inherent symmetrical disposition of functional groups around a pivotal hydroxyl group such as illustrated in Scheme 1, it is possible to capitalize on their manipulation and to produce stereochemically diverse patterns of substitution. Thus, manipulation of 3 gave the epoxide 11, mp 62-63°, $[\alpha]_D -9°$ (Scheme 3). Application of the sequence described above for 5, gave the chiral replicated butenolide 12, $[\alpha]_D -48.4°$ in high overall yield. Conjugate addition led to the lactone 13, $[\alpha]_D +12.2°$, which when reduced and protected as before, gave the acyclic derivative 14 in high overall yield. Reductive removal of the secondary hydroxyl group gave the (3R,5R)-dimethyl-1,7-heptanediol derivative 15, $[\alpha]_D +3.1°$, which was easily distinguishable from the diastereoisomeric 10 by high field ¹H n.m.r. spectroscopy.



a. BH₃.Me₂S, THF, 18h, (quant.); b. MsCl, pyr., DMAP, 2h; c. $(n-Bu)_4N^{+}F^{-}$, THF, 30 min., (71%, 2 steps); d. PhSCH₂CO₂H, LiN(TMS)₂, THF, 0° to 25°; add epoxide, 18h; e. ethyl (dimethylaminoethyl) carbodiimide HCl, DMAP, Et₂O, (84%, 2 steps); f. mCPBA, CH₂Cl₂, -23°, 30 min.; g. toluene, CaCO₃, reflux, 45 min., (92%, 2 steps); h. Me₂CuLi, Et₂O, 45 min.; i.LAH, Et₂O, 45 min., 93%; j. t-butyldiphenylsilyl chloride, CH₂Cl₂, pyr., DMAP, 18h, (98%); k. MsCl, pyr., CH₂Cl₂; then repeat i, (71%, 2 steps).

Finally, access to acyclic chains possessing a 1,4-C-methyl substitution pattern was easily accomplished by a combination of template-controlled lactone enolate alkylation,¹ followed by conjugated addition on a replicated butenolide as shown in Scheme 4. Thus, alkylation of 16 followed by transformation into an acyclic derivative as previously described,¹ then mesylation and treatment with fluoride ion gave the epoxide 17,

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 $[a]_D$ +4.1° in good overall yield. Chain-extension and transformation into the chiral butenolide 18, $[\alpha]_D$ -56.7°, followed by conjugate addition led to the chiral lactone 19, $[\alpha]_D$ +35.7° in 95% yield. Reduction, followed by silylation of the primary hydroxyl group, then gave the (2R,4R,5S)-2,6-dimethyl-1,4,7-heptanetriol derivative 20 in high yield, $[\alpha]_{D}$ +19.3°.



a. BH3.Me2S, THF, RT, 18h (~quant.); b. Ph3CC1, pyr., DMAP, RT, 18h, (94%); c. CH3SO2C1, pyr., RT, 2h; d. Bu4N*F*, THF, RT, 1h, (69%, 2 steps); e. PhSCH2CO2H, L1N(TMS)2, THF, 0°, then 25°; add epoxide, 18h; f. ethyl(dimethylaminoethyl)carbodiimide HC1, DMAP, ether, RT, 30 min., (87%, 2 steps); g. m-CPBA, CH₂Cl₂, -23°, 30 min.; h. toluene, CaCO₃, reflux, 45 min., (92%, 2 steps); i. Me₂CuLi, Et₂O, -23°, 30 min., (95%); j. LAH, Et₂O, RT, 1h, (91%).

It should be noted that several of the intermediates and/or acyclic products prepared by this strategy, have stereochemical plurality by virtue of the differentiation of the protected hydroxyl groups. Thus, products possessing inherent planes of symmetry (such as in 9,10 can be used as optically active precursors to segments of natural products with substitution patterns that are identical to, or enantiomeric with the precursors in ques-The same reasoning applies to products such as 15, where chain-elongation or shorttion. ening can be effected from either extremity at will to give diastereoisomeric compounds. Acknowledgements. We thank the National Engineering and Scientific Council of Canada and Le Ministère de l'éducation du Québec for generous financial support, and Mrs. A. Glamyan and P. Smith for technical assistance.

References

- 1. S. Hanessian, P.J. Murray and S.P. Sahoo, preceding publication.
- See for example, the ionophore antibiotic ionomycin, B. Toeplitz, A.T. Cohen, P.T. 2. Funke, W.L. Parker and J.Z. Gougoutas, J. Am. Chem. Soc., 101, 3344 (1979); A-23187, M.O. Chaney, P.V. DeMarco, N.D. Jones and J.L. Occolowitz, J. Am. Chem. Soc., <u>96</u>, 1932 (1974); see also the synthetic ACE inhibitor, N. Gruenfeld, J.L. Stanton, A.M. Yuan, F.H. Ebetino, L.J. Browne, C. Grade and C.F. Huebner, J. Med. Chem., 26, 1277 (1983).
- See for example, C.H. Heathcock, E.T. Jarvi and T. Rosen, Tetrahedron Lett., 25, 243 (1984); T. Nakata, M. Fukui, H. Ohtsuka and T. Oishi, Tetrahedron Lett., 24, 2661 (1983); C.S. Chen, Y. Fujimoto and C.J. Sih J. Am. Chem. Soc., 103, 3580 (1981); D.A. 3. Evans, C.E. Sacks, W.A. Kleschick and T.R. Taber, J. Am. Chem. Soc., 101, 6789 (1979); P.A. Grieco, E. Williams, H. Tanaka and S. Gilman, J. Org. Chem., 45, 3539 (1980);
- See for example, J.P. Vigneron, R. Méric, M. Larchevêque, A. Debal, G. Kunetch, P. Zagatti and M. Gallois, Tetrahedron Lett., 23, 5051 (1982); K. Tamioka, T. Ishiguro and 4. K. Koga, Tetrahedron Lett., 25, 2891 (1984), and references cited therein.
- Optical rotations were measured in chloroform. 5.
- J.P. Vigneron et al., Tetrahedron, <u>30</u>, 3521 (1984); M. Taniguchi, K. Koga and S. Yamada, Tetrahedron, <u>30</u>, 3547 (1970); see also ref. 1. P. Camps, J. Cardellach, J. Font,R.M. Ortuno and U. Ponsati,Tetrahedron,<u>38</u>,2395 (1982). 6.
- 7.
- K. Iwai, M. Iwai, H. Kosugi and H. Uda, Chem. Lett., 385 (1974). 8.

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